



Inventor: DiFoggio
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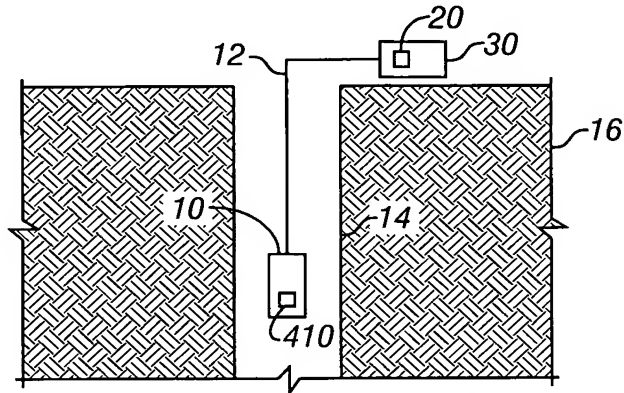


FIG. 1

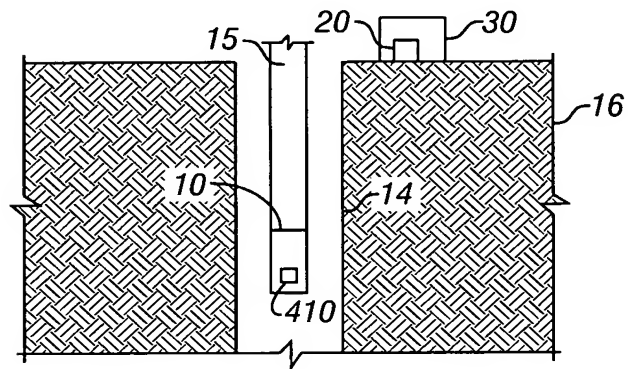


FIG. 2

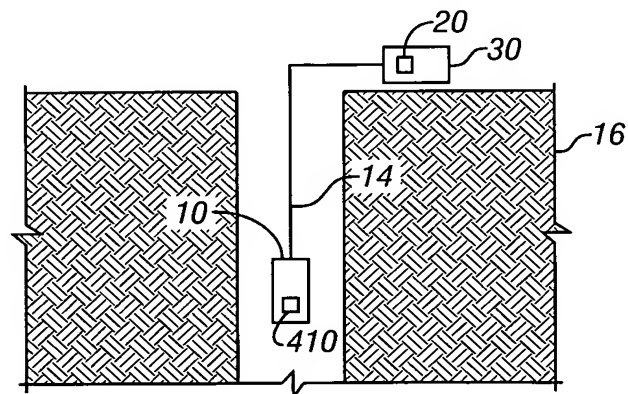


FIG. 3

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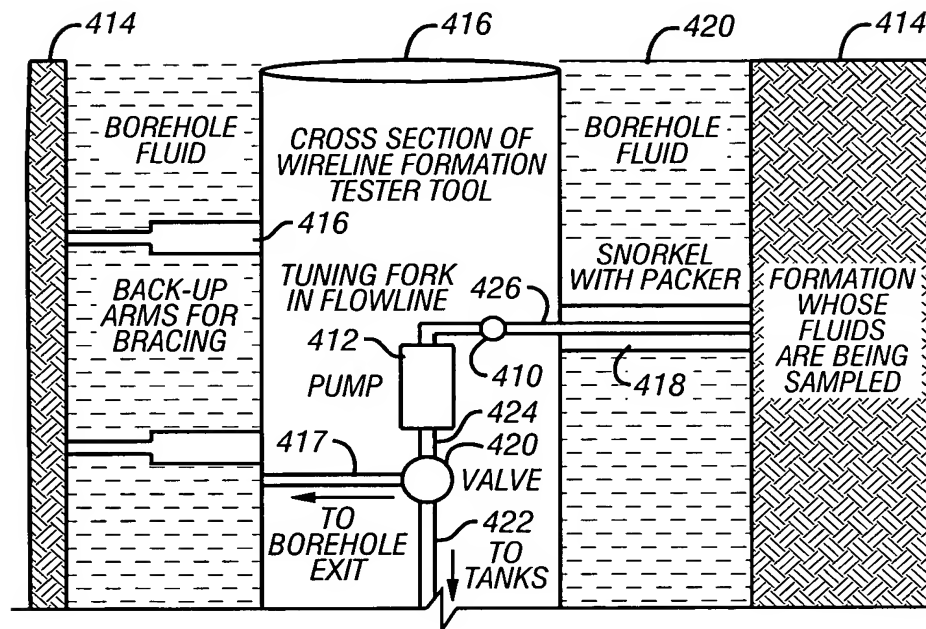


FIG. 4

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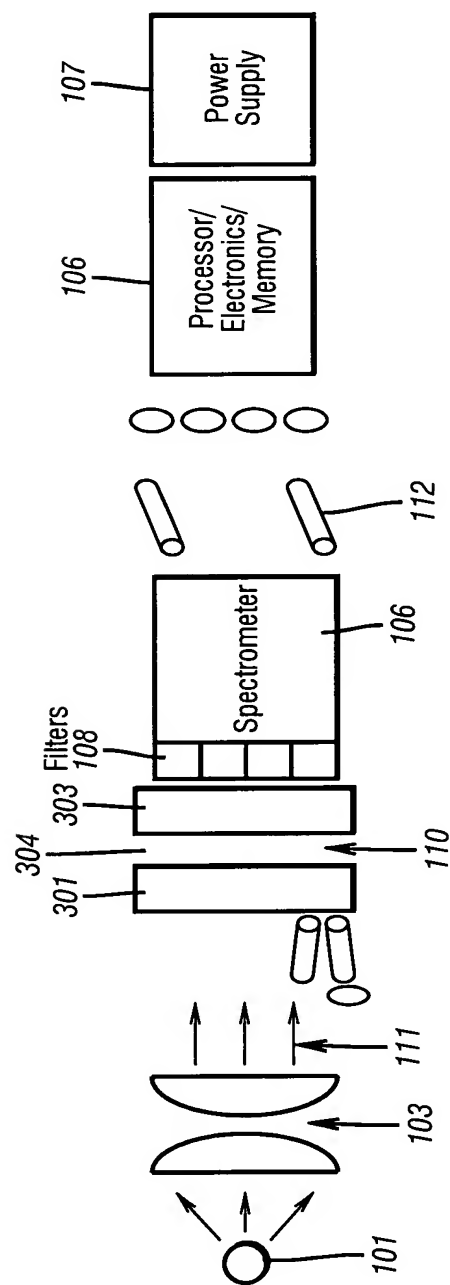


FIG. 5

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Equations Correlating Weight Fraction Methane in Mixtures of Crude Oil and Methane to Optical Absorbance and Temperature

Methane Weight Fraction = METHWTF = B0 + B1 * Var1 + B2 * Var2 + B3 * Var3 + B4 * Var4 ...

Regression Summary for Dependent Variable: METHWTF
 R = .98093203 R² = .96222765 Adjusted R² = .96151158
 F(4,211) = 1343.8 p<0.0000 Std. Error of estimate: .04992

B

0.06514 = B0 = Intercept
 11.1756 = B1
 0.00087 = B2
 -2.66167 = B3
 2.63245 = B4

Regression Summary for Dependent Variable: METHWTF
 R = .98190316 R² = .96413381 Adjusted R² = .96327986
 F(5,210) = 1129.0 p<0.0000 Std. Error of estimate: .04876

B

0.03143 = B0 = Intercept
 2.53111 = B1
 -2.55766 = B2
 11.9135 = B3
 0.0019 = B4
 -6.2E-06 = B5

SQ70-82 = SQUARE(Absorbance_at_1670_nm - Absorbance_1682_nm)
 SRSA1670 = SQRT(Absorbance_at_1670_nm)
 SRSA1682 = SQRT(Absorbance_at_1682_nm)
 TEMP_C = Temperature in Degrees Centigrade
 TEMP_SQR = Square of Temperature in Degrees C

Equation for Density of Methane [g/cc] as a Function of Pressure and Temperature from 100- 30,000 psia and 75-200 C is fitted by
 Adj. R² = .99911359
Equation for Optical Absorbance per mm of Methane as a Function of Density and Wavelength at 11 nm FWHM, Center λ range of 1668-1684 nm, for 100-30,000 psia and 75-200 C, is fitted by
 Adj. R² = .94145159

	B	
P	2.771E-03 = Intercept	B
P ²	2.480E-05	-19.9061 = Intercept
P ³	-1.120E-09 for Pressure in psi	Methane Density
T ²	1.808E-14	WaveNumber/1000
(P/T)	-1.308E-07 for Temperature in C	3.3326
(P/T) ²	1.455E-03	
(P/T) ³	-4.922E-06	
	5.934E-09	

Equations Relating Gas Oil Ratio, GOR, to Weight Fraction of Methane, f_M, and Stock Tank Density, ρ_O, of Oil

1 bbl = 0.159 m³ = 5.615 cu ft = 42 U.S. gal
 1 Standard Cubic Foot (SCF) or Methane Gas at 14.7 psia & 60°F is 0.042358 lbs = 19.21327 grams.
 Density of Methane at 60°F and 14.7 psia is 0.0006787 g/cc = 0.04258 lbm/ft³
 Letting V = Volume, W = Weight, ρ = Density, and using subscripts M for Methane and O for Oil,

$$GOR = V_{Methane} [SCR] / V_{Oil} [bbls] = \{W_M / (19.21 \text{ g/SCF})\} / \{(W_O / \rho_O) (1 \text{ bbl}/158.983 \text{ cc})\}$$

 Letting f_M = Weight Fraction of Methane,

$$GOR = 8274.62 \rho_O / (1/f_M - 1)$$

$$f_M = W_M / (W_M + W_O) = \rho_M V_M / (\rho_M V_M + \rho_O V_O) \text{ so } W_O = W_M / (1/f_M - 1) \text{ which substitutes into above.}$$

$$f_M = 1 / (1 + 8274.62 * \rho_O / GOR) \text{ where } W_G \text{ and } W_O \text{ are in grams, } \rho_O \text{ is in g/cc, and } f_M = \text{Wt. Frac. of Methane}$$

FIG. 6

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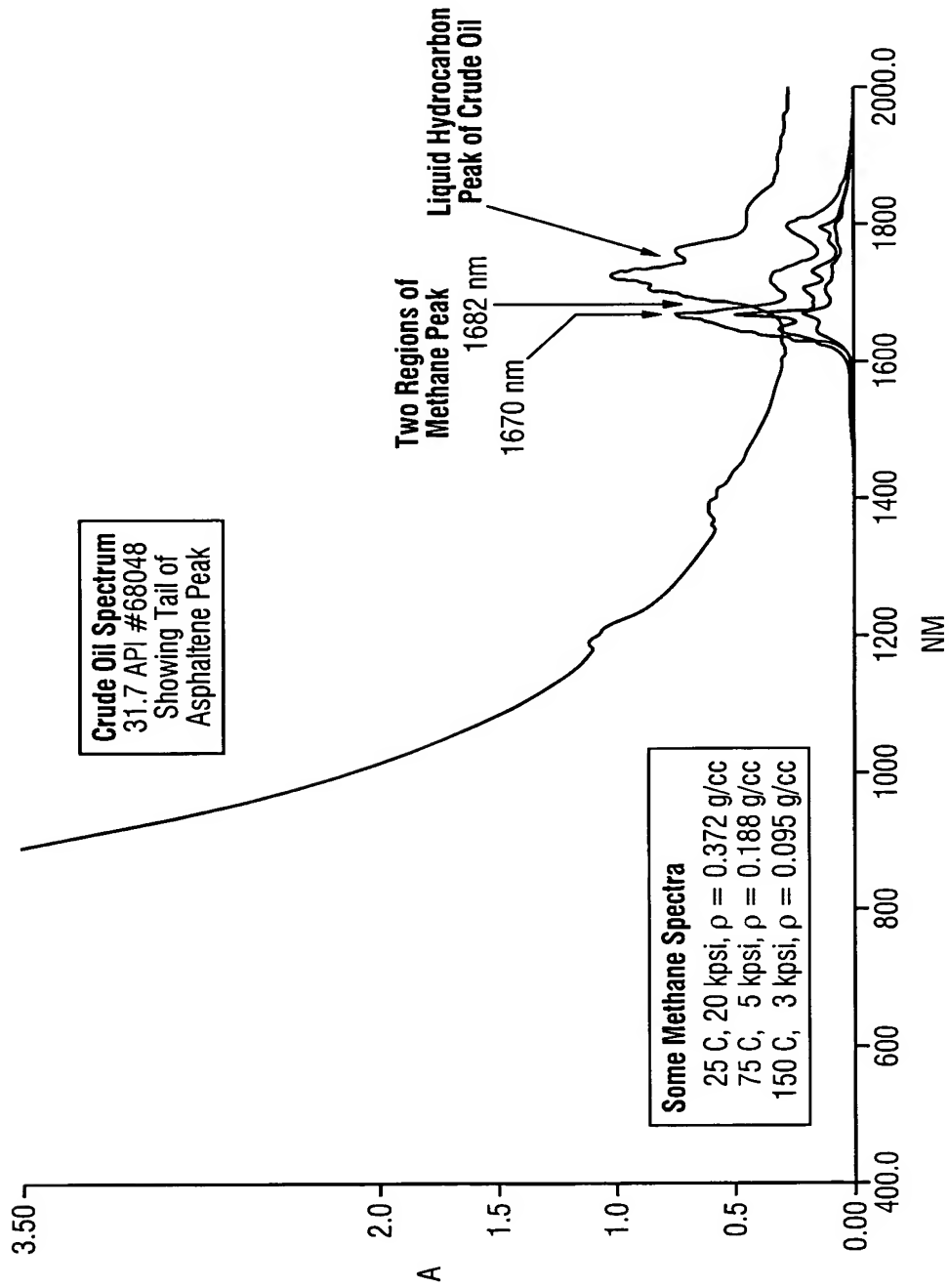


FIG. 7

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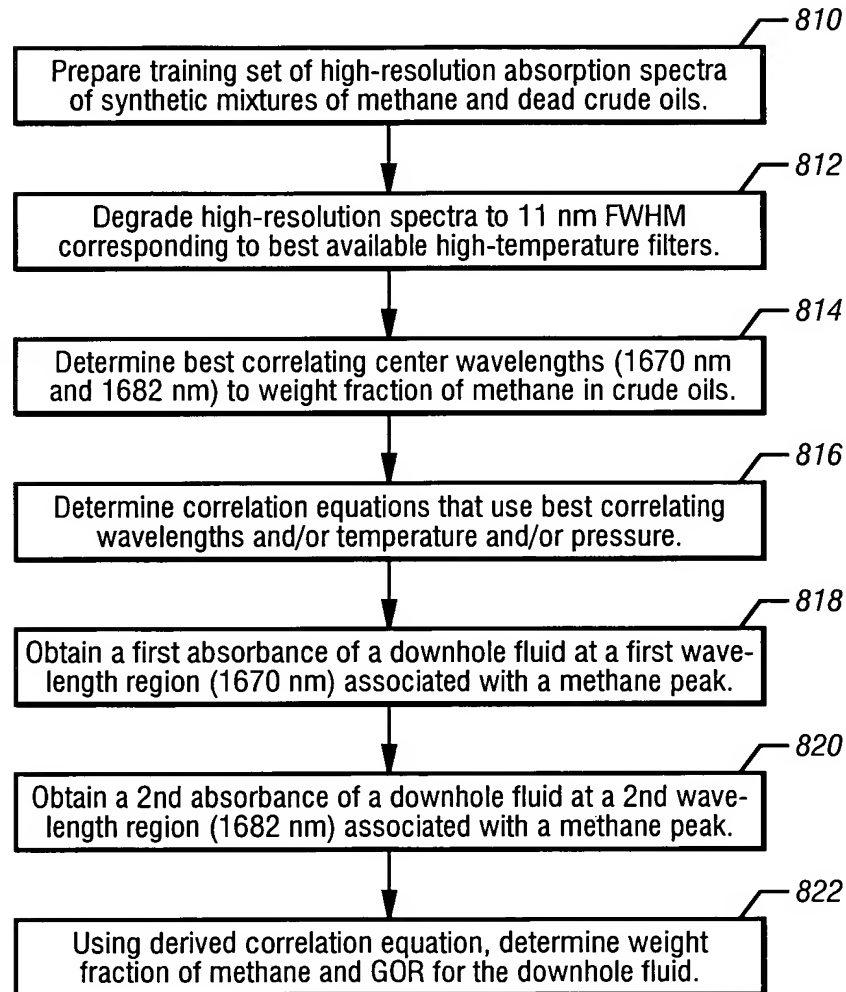


FIG. 8